


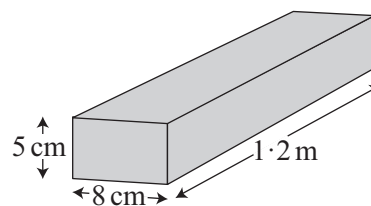
TASK 3.10

Use  to help you work out the questions below.

1. A solid weighs 450 g and has a volume of 50 cm^3 . Find the density of this solid.
2. A liquid has a density of 2 g/cm^2 . How much does the liquid weigh if its volume is 240 cm^3 ?
3. A metal bar has a density of 12 g/cm^3 and a mass of 360 g. Find the volume of the metal bar.
4. Copy and complete this table.

density (g/cm^3)	mass (g)	volume (cm^3)
7		90
	240	60
8	152	
	42	0.5
13	585	
1.5		140

5. Gold has a density of 19.3 g/cm^3 . A gold ring has a volume of 1.1 cm^3 . Find the mass of the gold ring.
6. A brass handle has a volume of 17 cm^3 and a mass of 139.4 g. Find the density of the brass.
7. Which has a greater volume — 102.6 g of lead with density 11.4 g/cm^3 or 78.85 g of steel with density 8.3 g/cm^3 ? Write down by how much.
8. The density of this metal bar is 7.4 g/cm^3 . Find the mass of this metal bar. Give your answer in kg. (Note the length is given in metres.)



9. A metal cube of length 0.2 m has a density of 8.3 g/cm^3 . A hole is bored through the cube with 485 cm^3 of metal being removed. What is the mass in kg of the remaining piece of metal?
10. A metal bar has 3 holes cut completely through its length. The cross-sectional area of each hole is $y \text{ cm}^2$. The density of the metal is 9 g/cm^3 . Find the mass of the remaining piece of metal, giving your answer in terms of x and y .

